

# Isolation, Purification and Characterization of Biodegradable Polymer Producing Bacteria *Pseudomonas pseudomallei*

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## ABSTRACT

Fifteen bacterial colonies were isolated, purified and preserved using synthetic enrichment medium. The isolates were then stained for poly-β-hydroxybutyric acid (PHB) with Sudan Black B stain. With the positive isolates, an attempt was made for the production, isolation and purification of biodegradable polymer, where 1.91% PHB related compounds were recorded from the proposed isolate QY<sub>1</sub>. On the basis of morphological, cultural, physiological and biochemical characteristics, the isolate QY<sub>1</sub> was identified as *Pseudomonas pseudomallei*. Finally, the *chloroform* extracts of biodegradable polymer was analysed by GCMS (Gas chromatography linked to Mass spectroscopy), where 28 different compounds were recorded. Among them, n-Hexadecanoic acid (stearic acid), oleic acid and phenyl isobutyrate are the major compounds.

**Key Words:** Isolation; Purification; Bacterial polymer; *Pseudomonas pseudomallei*

## INTRODUCTION

Biodegradable polymer plays a predominant role in the functioning of as biodegradable plastic due to their potentially hydrolysable ester bonds. This polymer family is made of two major groups- aliphatic and aromatic. Polyhydroxyalkanoate (PHAs) are aliphatic polymer naturally produced via a microbial process on sugar-based medium, where they act as carbon and energy storage material in bacteria. They were first biodegradable polymers to be utilized in plastics. The two main members of the PHA family are polyhydroxybutyrate (PHB) and polyhydroxyvalerate (PHV). PHB is accumulated inside a variety of microorganisms under appropriate conditions such as limitation of nitrogen, calcium, magnesium, iron or essential vitamins. PHB has also been found in numerous heterotrophic and autotrophic aerobic bacteria, photosynthetic anaerobic bacteria (Dawes & Senior, 1973), gliding bacteria (Prinsheim & Wiessner, 1963), actinomycetes (Kannan & Rehacek, 1970), cyanobacteria (Carr, 1966) and many other prokaryotes. The main advantage of this type of polymer is that, since, they are of biological origin, they degrade naturally and completely to CO<sub>2</sub> and H<sub>2</sub>O under natural environment by the enzymatic activities of microbes. Keeping all the points in mind, present study has been taken to isolate, purify, characterize and analyze the PHB producing microbes.

## MATERIALS AND METHODS

**Screening of PHB producing bacteria.** Fifteen bacterial colonies were isolated, purified and preserved using enrichment and nutrient agar medium. The isolates were

screened for PHB by staining with Sudan black B stain (0.3 in 70% alcohol) (Smibert & Krieg, 1981) and observed under microscope (X100x). The selected isolates were then identified on the basis of their morphological, cultural, physiological and biochemical characteristics.

**Optimization of culture medium and conditions.** To observe the effects of culture conditions for maximum bacterial polymer production, cultures were incubated at different incubation period (4, 8, 16, 24, 30, 36, 40 and 48 h) at 30°C.

The production of bacterial polymer under different carbon and nitrogen were also studied using liquid synthetic medium [Modified Okamoto Medium (MOM), described by Maeda, 2000] as basal medium. (Sodium acetate, Sodium malate, Sodium pyruvate, Sodium succinate)/ Glucose/ Fructose or Methanol were used as carbon source, whereas Ammonium sulfate, Asparagine and Yeast extract were tested for their ability to utilize nitrogen source. Biodegradable polymer production was also carried out in presence or absence of Vitamin solution (Thiamine, Nicotinic acid, p-Aminobenzoic acid, Cyanocobalamine, Vitamine-B<sub>6</sub> and distilled water, using MOM as basal medium).

**Cell cultivation.** For large-scale growth, 24 h old culture was prepared in nutrient broth medium at 30°C and transferred to 500 mL of nutrient broth in a wide-necked 1 L culture flask, incubated at 30°C with continuous gentle shaking (20 strokes/min).

**Biodegradable polymer accumulation in cells.** After 24 h of cultivation period, cells were harvested by centrifugation at 8000 rpm at 4°C for 12 min, washed aseptically with sterile distilled water and resuspended into 1 L culture bottle containing 500 mL of biodegradable polymer production

medium, which consist of 0.25 g of  $\text{KH}_2\text{PO}_4$ , 0.25 g of  $\text{K}_2\text{HPO}_4$ , 0.5 g of  $(\text{NH}_4)_2\text{SO}_4$ , 0.1 g of  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.1 g of  $\text{NaCl}$ , 0.02 g of  $\text{CaCl}_2 \cdot 7\text{H}_2\text{O}$  and 0.5 mL of solution 1 (2.0 g of EDTA, 2.0 g of  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.1 g of  $\text{H}_3\text{BO}_3$ , 0.1 g of  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ , 0.1 g of  $\text{ZnCl}_2$ , 0.1 g of  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ , 0.020 g of  $\text{Na}_2\text{MoO}_4$ , 0.020  $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ , 0.01 of  $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ , 0.010g of  $\text{Na}_2\text{SeO}_3$  and 1000 mL distilled water) and 500 mL of distilled water with the addition of different nitrogen and carbon source at a final concentration of 0.1%. Cells were then incubated with gentle shaking (20 strokes/min) at  $30^\circ\text{C}$  for 48 h.

**Harvesting of cells.** After incubation, cells were harvested by centrifugation at 8000 rpm for 12 min, washed in sterile water and recentrifuged similarly. Pellets were collected aseptically, dried at  $60^\circ\text{C}$  until constant weight and weight was measured.

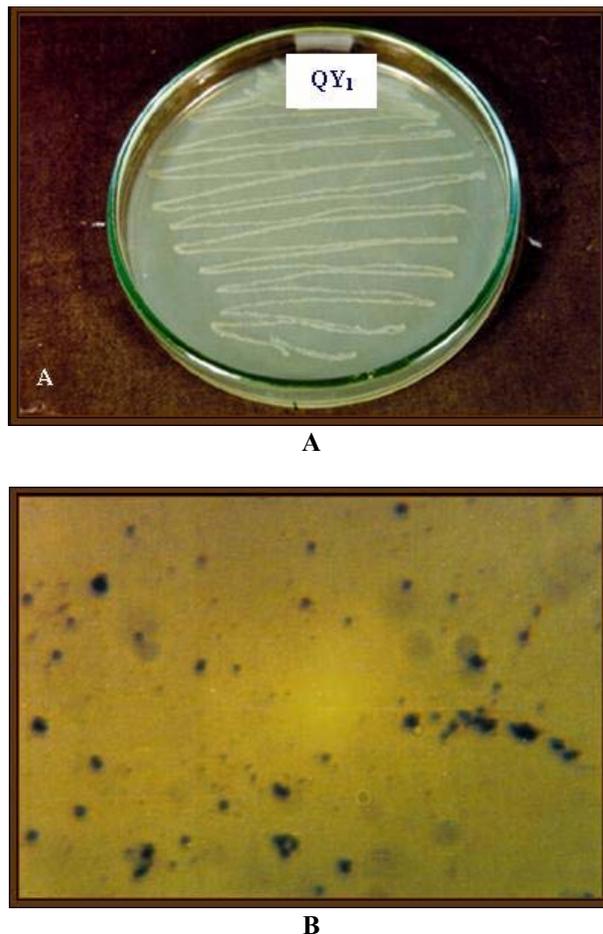
**Biodegradable polymer extraction.** Allumina was used to crush the cells with mortar and pestle. From the white powdery mass of the crushed cells, the polymer was extracted with chloroform ( $\text{CHCl}_3$ ). The chloroform mixtures were vigorously shaken, left for 1 h; the clear solution was carefully decanted into another clean test tube and evaporated. The powdery mass along the test tube wall was collected. Dry weight of the biodegradable polymer and percentage (w/w) of it against cell dry weight was measured.

**GCMS analysis.** The GCMS analysis was carried out in Bangladesh Council of Science and Industrial Research (BCSIR) Laboratories, Chittagong using electron impact ionization (EI) method on GC-17A gas chromatograph (Shimadzu, Japan) coupled to way GCM-QP 5050A mass spectrometer (Shimadzu); fused silica capillary column, 30 m x 0.25 mm i.e., coated with DB-5, 0.25  $\mu\text{m}$  film thickness; column temperature  $100\text{--}280^\circ\text{C}$  at the rate of  $4^\circ\text{C}/\text{min}$ ., injection port temperature  $250^\circ\text{C}$ ., constant pressure of carrier gas(helium)100 kPa, flow rate (mL/min) 20, acquisition parameters full scan, scan range 60 to 550 amu, searched library NIST 127 and 147, Shimadzu corporation, sample dissolved in chloroform.

## RESULTS AND DISCUSSION

Among the fifteen isolates, ten showed PHB positive reaction after staining with Sudan Black B stain. Out of 10 positive isolates,  $\text{QY}_1$  was finally selected for further studies. Isolate  $\text{QY}_1$  was then studied for morphological, cultural and biochemical characteristics (Fig. 1: A, B) and compared with the standard description of the Bergey's Manual of Determinative Bacteriology, 8<sup>th</sup> Ed. (Buchanan & Gibbons, 1974). The bacterial isolate  $\text{QY}_1$  was found to belong to the genus *Pseudomonas* and was provisionally identified as *Pseudomonas pseudomallei* (Whitmore) Haynes (1957). Many workers reported the ability of biodegradable polymer production by various members of the genera *Pseudomonas* (Anthony, 1976; Smet *et al.*, 1983; Suzuki *et al.*, 1986; Taylor *et al.*, 1989).

**Fig. 1. Photographs showing (A) Colonies on NA plate (B) Sudan Black staining of the isolate  $\text{QY}_1$ .**



**Medium composition and culture conditions of the culture.** Table I shows that maximum biomass and biodegradable polymers were produced when fructose was used as the carbon source and yeast extract used as the nitrogen source. Anderson and Dawes (1990) and Brauneegg *et al.* (1978) showed accumulation of PHB by *Alcaligenes faecalis* using fructose as the carbon source. Use of yeast extract as the nitrogen source has been reported by Fukui (1976), Nishimura *et al.* (1978) and Fernandez-Castillo *et al.* (1986). No considerable effect of vitamins on biomass and biodegradable polymer production was observed.

**Biodegradable polymer production.** An attempt was made for the production, isolation and purification of biodegradable polymer of  $\text{QY}_1$ . Biodegradable polymer was extracted from the crushed cells with chloroform, then dried and measured. The ratio of the biodegradable polymer was determined and recorded as 1.91% in case of the isolate  $\text{QY}_1$  (Table II). Similarly, biodegradable polymer extraction with chloroform from bacterial biomass was also reported by many other researchers (Schlegel *et al.*, 1961; Wallen & Rohwedder, 1974; Brauneegg, 1978; Findlay & White, 1983;

**Table I. Effect of carbon, nitrogen and vitamin on the Biomass and Biodegradable Polyester production of isolate QY<sub>1</sub> (*Pseudomonas pseudomallei*)**

Carbon sources	Nitrogen sources			
	Ammonium sulfate	Asparagine	Yeast extract	Yeast extract + Ammonium sulfate
Sodium acetate+				
Sodium malate+	Vit + 0.02	0.03	0.03	0.00
Sodium pyruvate +	Vit - 0.03	0.00	0.01	0.05
Sodium succinate				
Glucose	Vit + 0.19	0.22	0.29	0.60
	Vit - 0.19	0.08	0.29	0.58
Fructose	Vit + 0.10	0.42	0.85	0.25
	Vit - 0.17	0.41	0.85	0.28
Methanol	Vit + 0.00	0.09	0.22	0.15
	Vit - 0.01	0.09	0.14	0.13

**Table II. Cell dry weight and Biodegradable Polymer content (%) of the isolate QY<sub>1</sub> (*Pseudomonas pseudomallei*)**

Total cell dry weight (g)	Biodegradable weight (g)	Biodegradable polyester content (%)
1.5657	0.03	1.91

Fernandez-Castillo *et al.*, 1985; Suzuki *et al.*, 1986; Anderson & Dawes, 1990; Anwar & Hakim, 2002).

**Analysis.** The chloroform extracts of biodegradable polymer were dried and analyzed by GCMS (Qp5050A). Table III shows the result of GCMS analysis, where twenty eight different biodegradable compounds were found from chloroform extract. The major compounds among the analyzed compounds were n-Hexadecanoic acid (Stearic acid), Oleic acid and Phenyl isobutyrate (Fig. 2). The n-hexadecanoic acid is an aliphatic polymer esters. This aliphatic biodegradable polyester family due to hydrolysable ester bonds was reported by Dawes (1988) and others (Anonymous, 2002).

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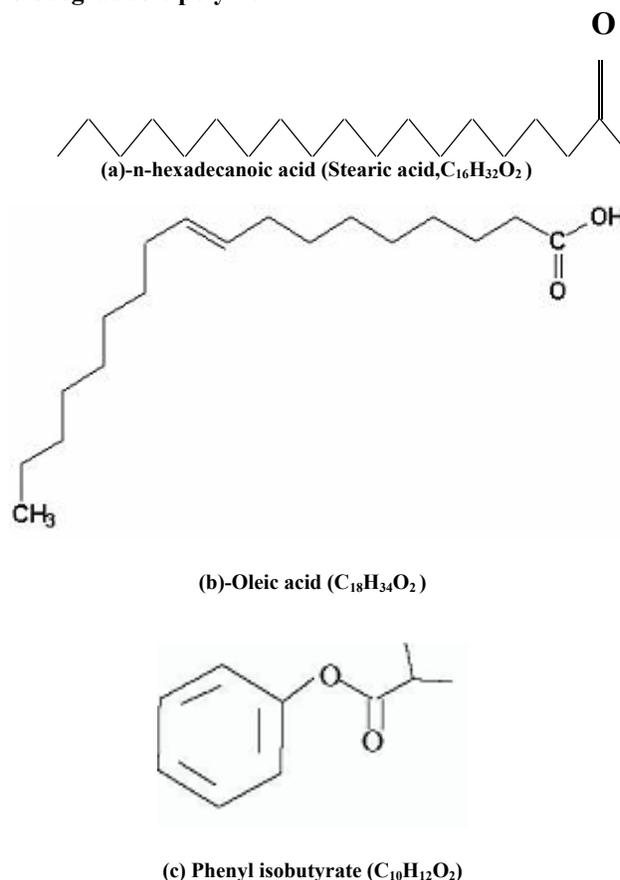
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**Fig. 2. The structure of (a)-n-hexadecanoic acid (Stearic acid), (b)-Oleic acid and (c)- Phenyl isobutyrate recovered from chloroform extraction of biodegradable polymer**



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**Table III. Percentage of different components with MW and MF present in Chloroform extract of biodegradable polymer-of isolate QY<sub>1</sub> (*Pseudomonas pseudomallei*)**

Sl. No.	Retention Time	% Total	Name of Compound	Molecular Formulae	Molecular Weight
1	14.205	0.17	n-Cetyl alcohol/Aldol	C <sub>16</sub> H <sub>34</sub> O	242
2	14.814	0.40	Tetradecanoic acid,	C <sub>14</sub> H <sub>28</sub> O <sub>2</sub>	228
3	15.395	1.46	Tetradecanoic acid,	C <sub>14</sub> H <sub>28</sub> O <sub>2</sub>	228
4	16.494	2.38	Pentadecanoic acid	C <sub>15</sub> H <sub>30</sub> O <sub>2</sub>	242
5	16.629	1.00	Pentadecanoic acid	C <sub>15</sub> H <sub>30</sub> O <sub>2</sub>	242
6	17.139	0.47	Hexadecanoic acid, ethyl ester	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	284
7	17.233	0.30	1,2-Benzenedicarboxylic acid/ Butyl Octylster	C <sub>20</sub> H <sub>30</sub> O <sub>4</sub>	334
8	19.067	3.04	Oleic acid 9-ctadecanoic acid (2).delta	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
9	19.620	59.68	n-Hexadecanoic acid(Stearic acid)	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	256
10	20.055	2.64	Hexadecanoic acid/ ethyl ester/Palmitic acid	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	284
11	20.473	0.19	Dodecanoic acid	C <sub>12</sub> H <sub>24</sub> O <sub>2</sub>	216
12	20.796	7.46	Oleic acid	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
13	20.942	0.39	2-t-Butyl-6-Chloroquino l-4-yl-2-Pyridyl Ketone	C <sub>19</sub> H <sub>17</sub> C <sub>12</sub> N <sub>2</sub> O	324
14	21.149	0.25	Ethyl Oleate	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	310
15	21.733	1.16	Oleic acid 9-ctadecanoic acid(2).delta	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
16	21.793	4.82	Oleic acid 9-ctadecanoic acid(2).delta	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
17	21.909	1.11	Oleic acid 9-ctadecanoic acid (2).delta	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
18	21.961	3.08	Octadecanoic acid	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	284
19	22.067	0.45	Oleic acid	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
20	22.108	0.21	Oleic acid	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
21	22.191	0.58	Oleic acid	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	282
22	22.256	0.67	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
23	22.652	0.61	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
24	23.860	1.17	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
25	24.631	0.50	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
26	24.802	4.59	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
27	26.961	0.34	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164
28	27.595	0.91	Phenyl isobutyrate	C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	164

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